

Shaded squares represent geometries which give non-polar molecules when all substituents *X* are identical.

<b>Total # of Groups of e<sup>-</sup></b>	<b>Electron Pair Geometry (Hybridization)</b>	<b>Approximate Bond Angle</b>	<b># of Bonding Directions (# of X)</b>	<b># of Lone Pairs (# of E)</b>	<b>Geometry Name (VSEPR class)</b>